ARTICLE OPEN Thermal stability and electronic and magnetic properties of atomically thin 2D transition metal oxides

Heleen van Gog¹, Wun-Fan Li¹, Changming Fang², Rik S. Koster¹, Marjolein Dijkstra¹ and Marijn van Huis¹

Two-dimensional (2D) transition metal oxides (TMOs) are an emerging class of nanomaterials. Using density functional theory and ab initio molecular dynamics (AIMD) simulations, we carried out a systematic study of atomically thin metal oxide phases with compositions MO, M_2O_3 , and MO_2 , for transition metal elements Sc, Ti, V, Cr, and Mn. We identified nine thermally stable structures that may be realized as free-standing nanosheets: hexagonal h-Sc₂O₃, h-V₂O₃, and h-Mn₂O₃; hexagonal t-VO, t-CrO, and t-MnO; and square sq-TiO, sq-VO, and sq-MnO. The t-MO phases are novel hexagonal structures which emerged naturally from phase transformations observed during AIMD simulations. The 2D TMOs were found to exhibit a wide range of remarkable electronic and magnetic properties, indicating that they are bright candidates for electronic and spintronic applications. Most exceptional in this regard is h-V₂O₃, that is the only phase that has been experimentally realized so far, and was found to be a ferromagnetic half-metal with Dirac-cone-like bands.

npj 2D Materials and Applications (2019)3:18; https://doi.org/10.1038/s41699-019-0100-z

INTRODUCTION

The discovery of graphene and its fascinating properties¹ has incited a whole new field of research dedicated to the exploration of two-dimensional (2D) nanomaterials. Using top-down approaches, where monolayers are isolated from layered bulk structures, and bottom-up approaches, where thin films are obtained through deposition of molecules on substrates, added to graphene is now a wide range of 2D materials consisting of elements other than carbon, such as *h*-BN, the mono-elemental 2D semiconductors silicene, phosphorene, and germanene, Mxenes, i.e., transition metal carbides, nitrides, and carbonitrides, and transition metal dichalcogenides (TMDs), such as MoS_2 .^{2,3} The exploration of 2D nanomaterials has only started little over a decade ago, but is rapidly progressing.

Bulk phase transition metal oxides (TMOs) are known to have unique properties originating from their itinerant and strongly correlated transition metal (TM) d electrons. This strong electron correlation, where the TM d electrons experience a strong Coulomb repulsion from one another, leads to physical phenomena such as the metal-insulator transition^{4,5} (MIT), where, upon tuning parameters such as temperature, pressure, and chemical composition, and often along with a structural and magnetic phase transition, the electronic structure of a system will undergo a MIT (for VO₂, for example, see refs. $^{6-8}$). In addition, TM *d* orbitals can be either fully or partially occupied, giving rise to multivalence states of TM atoms and allowing numerous stoichiometries; vanadium oxide, for example, existing in the four principle phases VO, V_2O_3 , VO₂, and V₂O₅, in the Wadsley phases⁹ V_nO_{2n+1} ($1 \le n \le 3$), and in the Magnéli phases¹⁰ V_nO_{2n-1} ($3 \le n \le 9$). A TMO can furthermore exist as various stable polymorphs, stoichiometrically identical but crystallographically different phases, that for MnO₂, for example, add up to a total number of six stable phases found so far¹¹ and may transform into one another with energy barriers as low as 0.2–0.3 eV/f.u.¹² Their diversity makes TMOs versatile in a wide range of applications, including energy storage, catalysis, molecular sieving, functional magnetic and optical materials, and biosensors.

With the popularity of 2D materials and the unique properties of bulk TMOs in mind, 2D TMOs are being actively investigated as well, and recently, the first thin film and nanosheet 2D TMOs have been realized experimentally. Limiting ourselves to the scope of our study, i.e., the early 3d transition metal (Sc, Ti, V, Cr, and Mn) oxides, and just naming a few works with no intention of a complete review: via chemical vapor deposition, Chang et al. prepared rutile TiO₂ and monoclinic VO₂ thin films on α -Al₂O₃ surfaces;¹³ with the same technique, epitaxial rutile and monoclinic VO₂ films were grown on a *r*-Al₂O₃ substrate by Makarevich et al.,¹⁴ monolayered (ML) V_2O_3 thin films have been synthesized on a Pd(111) surface by Surnev at al.^{15,16} Exfoliation of a layered manganese oxide by Omomo et al. resulted in ML nanosheets of MnO₂ in the T phase.¹⁷ A generalized approach to molecular selfassembly synthesis of ultrathin 2D nanosheets of transition metal oxides was developed by Sun et al., and resulted in the successful synthesis of, among others, ~3.3 -nm-thick TiO₂ nanosheets containing 4-5 MLs of anatase TiO2.18 The preparation of 2D TMOs on temporary, water-soluble salt templates resulted in ~1 nm-thick cubic MnO nanolayers and was reported by Xiao et al.¹⁹

Comprehensive computational studies using density functional theory (DFT) and the more advanced quasiparticle *GW* approach²⁰ (G: Green's function; *W*: screened Coulomb interaction) on monolayered MX₂ transition metal dichalcogenides and oxides (TMDs and TMOs) in the T and H phases were performed by Ataca et al. and Rasmussen et al.,^{21,22} and included ScO₂ and MnO₂, which are stable in both phases, TiO₂, which is unstable in either phase, and VO₂ and CrO₂, which only have a stable H phase.²¹ The monolayered V₂O₃ synthesized on a Pd(111) surface, mentioned in the previous paragraph, was characterized by DFT simulations by

Received: 11 January 2019 Accepted: 3 April 2019 Published online: 23 April 2019



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Kresse et al.²³ DFT calculations by Vittadini et al. on 1–6 monolayer (ML) thick films derived from the main low-index surfaces of anatase TiO₂ revealed two anomalously stable films that are a restructuring of anatase TiO₂ into a lepidocrocite-TiO₂ nanosheet (derived from the 2 ML (001) film) and a novel TiO₂ phase (derived from the 4 ML (101) film).²⁴ By DFT+ U^{25} (Hubbard *U* parameter) and hybrid HSE06²⁶ (Heyd–Scuseria–Ernzerhof) calculations, Kan et al. identified the experimentally realized monolayered T-phase MnO₂ nanosheets mentioned in the paragraph above to be an intrinsic ferromagnetic 2D material.²⁷ Graphitic, hexagonal MnO layers were further predicted by means of DFT+U by Kan et al. to be more stable than the corresponding wurtzite *wz*-MnO(0001) surface, when the number of layers is less than four.²⁸ Along this line, Wang et al. performed DFT calculations and suggested that semiconductor *wz* structures transform into planar hexagonal structures upon a planar stretching force.²⁹

To the best of our knowledge, a systematic first-principles' study of single-atomically thin 2D TMO structures has not been conducted yet. Inspired by the successful synthesis of 2D hexagonal V₂O₃ on a substrate and the predicted stability of 2D hexagonal MnO, we designed four different 2D TMO phases: hexagonal MO (h-MO) and square MO (sq-MO), which both have an equivalent bulk phase, namely wurtzite MO (wz-MO) and rocksalt MO (rs-MO), that can be built up directly from the 2D counterparts; and hexagonal M_2O_3 (h-M_2O_3) and square MO_2 (sq-MO₂), for which no direct 2D bulk connection exists. Using planewave DFT and hybrid HSE06 calculations, we systematically disclosed the energetic and thermal stability, and novel electronic and magnetic properties, of these atomically thin 2D TMOs for the five early 3d transition metals Sc, Ti, V, Cr, and Mn. Except for h-V₂O₃ and *h*-MnO, all these structures are at this moment undiscovered.

RESULTS AND DISCUSSION

Computational methodology

The four planar *h*-MO, *sq*-MO, *h*-M₂O₃, and *sq*-MO input geometries considered in this study for the five TMs, Sc, Ti, V, Cr, and Mn, are shown in Fig. 1. For each of the 2D TMO input geometries, non-magnetic (NM), ferromagnetic (FM), and various anti-ferromagnetic (AFM) orderings of the TM atoms were investigated, including for *h*-MO, the ferrimagnetic (FiM) configuration that is the ground-state magnetic ordering of half-hydrated graphene (graphone).³⁰ The number of (planar) structural-magnetic input configurations adds up to 20 for *h*-MO, 20 for *sq*-MO, 15 for *h*-M₂O₃, and 20 for *sq*-MO₂. Inspired by the experimentally synthesized buckled structure of

 $h-V_2O_3$,^{15,16,23} we also examined the stability of buckled 2D TMO phases, rerunning all simulations with buckled instead of planar input geometries, with the O-sublayer shifted 0.5 Å away from the TM-sublayer, bringing the total number of structural and magnetic input configurations considered in this study for the *h*-MO, *sq*-MO, *h*-M₂O₃, and *sq*-MO₂ phases to 150.

The DFT calculations were performed using the first principles' Vienna Ab initio Simulation Package^{31–33} (VASP). Within the projector-augmented wave (PAW) framework,^{34,35} the exchange and correlation energy terms were described using the Perdew–Burke–Ernzerhof (PBE) generalized gradient approximation (GGA) functional.^{36,37} Computational details are given in the Methods section. To obtain ground-state geometries and magnetic arrangements, the simulation cells were structurally and magnetically optimized, relaxing lattice parameters, atomic positions, and magnetic moments. To study the charges of the atoms, a Bader charge analysis³⁸ was carried out using the approach by Henkelman et al.^{39,40}

The enormous amount of calculations involved in our study prevented us from relaxing the structures with the more advanced Hartree-Fock (HF)/DFT hybrid Heyd-Scuseria-Ernzerhof functional²⁶ (HSE06). Previous studies have shown, however, that using HSE for 2D materials such as graphene halides⁴¹ and MoS_2^{-1} results in geometries that are almost identical to PBE-relaxed geometries. In addition, studies of bulk TMOs have found, at the same time, that, while giving very similar results for structural relaxation and relative stability and indicating that the nonlocal HF exchange has little effect on the structure and total energy of strongly correlated systems, the description of the electronic structure of such a system, in particular of its d states, greatly improves when using a hybrid functional.^{11,43–47} To obtain a more accurate description of the electronic structure of our 2D TMOs, we therefore performed HSE06 calculations on the PBE-relaxed structures that were found to be the 2D TMOs' structural and magnetic ground-state configurations. Employing the popular and computationally less expensive DFT + U method was not considered in this study, as we are investigating a wide range of elements, and various physical properties such as energy, geometry, electronic structure, and magnetism, and the DFT + Uapproach depends on the Hubbard U parameter, the value of which differs for different elements and different properties, and may lead to unwanted inconsistencies in our results.⁴

To assess the thermal stability of the 2D TMOs, we performed ab initio molecular dynamics (AIMD) simulations for each of the 20 structural and magnetic ground-state configurations, using 72-atom 6×6 supercells for the *h*-MO phases, 100-atom 5×5 supercells for the *sq*-MO phases, 80-atom 4×4 supercells for the

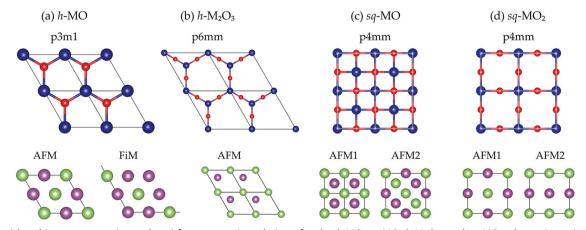


Fig. 1 Considered input geometries and anti-ferromagnetic orderings for the h-MO, sq-MO, h-M₂O₃, and sq-MO₂ phases investigated in this study. Dark blue and red spheres denote TM and O atoms, respectively. Light green and purple spheres denote spin-up and spin-down TM atoms, respectively. Boundaries of simulation and unit cells are indicated with black or white solid lines

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Table 1.	Summary of the structural,	electronic, and magnetic properties	s, and thermal stability,	of the structural and m	agnetic ground-state
configura	ations for the 2D TMOs iden	tified in this study			

2D TMO	a (Å)	θ (°)	E _f (eV/atom)	d _ь (Å)	EGS	MGS	E _{g,↑} (eV)	E _{g,↓} (eV)	т _{тм} (μ _в)	m _Ο (μ _Β)	q _{в,тм} (е)	q _{в,0} (е)	Thermally stable?
t-VO (a)	5.423	119.5	-1.335	0.44	SC	AFM2	1.16	1.16	±2.27	±0.04	1.09	-1.09	Yes
t-CrO (a)	5.435	119.8	-1.174	_	SC	AFM2	0.60	0.60	±3.36	±0.06	1.09	-1.09	Yes
t-MnO (b)	a ≠ bª	118.5	-1.208	_	SC	AFM1	1.69	1.69	±4.42	±0.07	1.30	-1.30	Yes
h-ScO	3.774	60.4	-2.367	0.53	SC	AFM	1.05	1.05	±0.21	±0.02	1.41	-1.41	No
<i>h</i> -TiO	3.092	60.0	-1.753	0.71	SC	NM	0.28	_	_	_	1.27	-1.27	No
h-VO	3.236	60.0	-1.261	0.39	SC	FM	0.55	6.77	2.86	0.14	1.29	-1.29	No
h-CrO	3.379	52.3	-1.038	_	М	AFM	_	_	±3.66	±0.08	1.32	-1.32	No
<i>h</i> -MnO	3.282	64.2	-1.137	_	SC	AFM	2.11	2.11	±4.61	±0.08	1.42	-1.42	No
sq-ScO	4.221	90.0	-2.470	_	М	NM	_	_	_	_	1.62	-1.62	No
sq-TiO	4.038	90.0	-1.942	_	М	AFM1	_	_	±1.22	0.00	1.50	-1.50	Yes
sq-VO	3.973	90.0	-1.430	_	SC	AFM1	3.12	3.12	±3.66	0.00	1.46	-1.46	Yes
sq-CrO	3.951	90.0	-1.250	0.22	М	AFM1	_	_	±2.61	0.00	1.45	-1.45	No
sq-MnO	4.059	90.0	-1.098	_	SM	AFM1	-0.04	-0.04	±4.64	0.00	1.44	-1.44	Yes
h-Sc ₂ O ₃	6.650	60.0	-2.713	_	SC	NM	4.35	_	_	—	1.99	-1.33	Yes
h-Ti₂O₃	6.301	60.0	-2.315	0.15	SC	AFM	3.23	3.23	±0.95	0.00	1.81	-1.21	No
$h-V_2O_3$	6.181	60.0	-1.821	_	HM	FM	—	6.24	2.10	-0.06	1.65	-1.10	Yes
h-Cr₂O₃	6.137	60.0	-1.394	_	SC	FM	0.88	5.73	2.93	0.05	1.65	-1.10	No
<i>h</i> -Mn ₂ O ₃	6.159	60.0	-1.225	_	SC	FiM	3.02	3.77	FiM ^b	0.05	1.44	-1.02	Yes
sq-ScO ₂	4.008	90.0	-1.980	_	HM	FM	5.21	_	-0.09	0.54	2.08	-1.04	No
sq-TiO ₂	3.746	90.0	-2.144	_	SC	NM	0.13	_	_	_	2.28	-1.14	No
sq-VO ₂	3.647	90.0	-1.681	_	SC	FM	0.65	1.93	1.12	-0.06	2.06	-1.03	No
sq-CrO ₂	3.580	90.0	-1.378	_	HM	FM	_	2.91	2.54	-0.27	1.93	-0.97	No
sq-MnO ₂	3.556	90.0	-1.114	_	НМ	AFM1	4.07	_	±3.48	0.50	1.93	-0.96	No

^aFor *t*-MnO (b), lattice parameters are: a = 5.710 Å and b = 5.447 Å

^bFor FiM *h*-Mn₂O₃, transition metal magnetic moments m_{TM} are: 4.55 μ_B and -2.69 μ_B

PBE-calculated: lattice parameters a and θ , formation energy E_{f_i} and buckling distance d_b . HSE06-calculated: electronic ground-state EGS, magnetic groundstate MGS, spin-up and spin-down electronic bandgaps $E_{g,1}$ and $E_{g,1}$, transition metal and oxygen magnetic moments m_{TM} and m_O , transition metal and oxygen Bader charges $q_{B,TM}$ and $q_{B,O'}$ and thermal stability. For the *t*-MOs, "(a)" and "(b)" refer to the configurations shown in Fig. 5a, b, respectively

 $h-M_2O_3$ phases, and 72-atom 5 × 5 supercells for each of the sq-MO₂ phases, except sq-MnO₂, which was simulated using a 108atom 6 × 6 supercell to accommodate its magnetic ordering. For each of the structures, with a time step of 1 fs, a 2-ps initial equilibration and a 5-ps subsequent canonical (NVT) ensemble simulation using a Nosé thermostat^{49–51} was carried out, at simulation temperatures of 300 and 700 K. After the AIMD simulations, AIMD supercell output configurations were relaxed at high accuracy at 0 K again, to see if the structures had remained topologically equivalent.

Energetic stability of the *h*-MO, sq-MO, h-M₂O₃, and sq-MO₂ phases

To assess the relative stability of the 2D TMO phases, we define their formation energy per atom, E_{f_1} relative to their elemental transition metal bulk phases TM and the paramagnetic oxygen molecule O_2 :

$$E_{f} = \frac{1}{1+x} \Big[E(MO_{x}) - E(TM) - \frac{x}{2} E(O_{2}) \Big],$$
(1)

where $E(MO_x)$ is the calculated total energy of a 2D TMO with x = 1, 1.5, or 2; E(TM) is the energy per atom of the TMs in their elemental phase; and $E(O_2)$ is the energy of an O_2 molecule in its paramagnetic ground state.

Table S1 in the Supplementary Information (SI) lists the calculated energies of lowest-energy and higher-energy magnetic and nonmagnetic configurations for all 2D TMO phases

considered in this study. This table includes *t*-MO phases, which were found from the AIMD simulations and which will be discussed separately below. A summary of the formation energy E_f of the most stable magnetic ground-state configurations can be found in Table 1. For the latter, a concave-hull plot is included as well (see Fig. 2). In general, a system will be unstable and decompose into its two neighboring compositions if its energy lies above the concave-hull line, i.e., the common tangent constructed connecting line. In the concave-hull plot for the 2D TMOs, for each of the phases the formation energy is located below the straight line connecting the two adjacent data points. This suggests that all considered 2D TMO phases are energetically stable at the respective stoichiometries.

Comparing the formation energies of the *h*-MO and *sq*-MO polymorphs, the *sq*-MO phase was found to be energetically more favorable than the *h*-MO phase for all considered TMs except Mn, *sq*-MnO being less stable than *h*-MnO by 39 meV/atom. The relative stability of the *sq*-MO phases over *h*-MO may be due to the bonding network, for which *sq*-MO, where each TM atom is coordinated by four oxygen atoms, resembles that of its 3D *rs* counterpart. In general, *rs* compounds have a stronger ionic nature than corresponding *wz* phases. The stronger ionicity of the TM–O bonds in *sq*-MOs is indeed evidenced by the Bader charge analysis of the 2D structures: as can be seen from Table 1, for all TMs except Mn, the TM atoms were found to have a Bader charge that is ~0.2*e* higher in the *sq*-MO phase than in the corresponding *h*-MO phase; for MnO, the Bader charge on the Mn atoms was

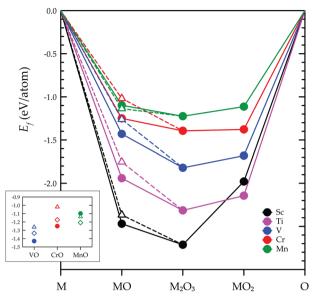


Fig. 2 Concave-hull plot of the formation energies E_f of the *h*-MO phases (triangle markers), and sq-MO, h-M₂O₃, and sq-MO₂ phases (circle markers). The inset shows the formation energies of the three *t*-MO phases (diamond markers) and their corresponding *h*-MO (triangle markers) and sq-MO (circle markers) phases. The formation energies are also listed in Table 1

found to be similar in magnitude in both phases. TM and O atom coordination numbers and TM-O bond lengths in the 2D TMOs (listed in Table S2) can be correlated with the calculated atomic Bader charges (listed in Table 1): In sq-MO, the coordination number of all atoms is 4, whereas for h-MO this number is 3. The larger number of bonds correlates with an increased ionicity for sq-MO structures, while at the same time the bond lengths in the latter are slightly larger than those in corresponding h-MO structures. Comparing MO, M₂O₃, and MO₂ structures, it can be observed that the Bader charge on the TM atoms is dominated by the difference in stoichiometry, and increases when going from MO to M_2O_3 to MO_2 . For the latter two structures, the bond lengths are considerably smaller than those in h-MO and sq-MO phases, by up to 10%, which is likely related to the fact that the O atoms have only two TM nearest neighbor atoms in M₂O₃ and MO₂.

For further analysis of the structural buckling, we define the buckling energy E_b as the energy difference between a d_b buckled structure and its planar counterpart:

$$E_b = E(\mathsf{TMO}_{\mathsf{buckled}}) - E(\mathsf{TMO}_{\mathsf{planar}}), \tag{2}$$

where a negative E_b indicates that a buckled structure is energetically more stable than the corresponding planar structure, and d_b is simply the perpendicular separation between the TMand O-sublayer.

Relaxation of the 150 input configurations for the *h*-MO, *sq*-MO, *h*-M₂O₃, and *sq*-MO₂ phases resulted in 24 stable and metastable buckled structures. Of these, 15 were found to have $|E_b| \le 20 \text{ meV}/$ atom and $d_b \ge 0.02 \text{ Å}$ (see Table S1), implying that these 2D TMO phases have a rather flat potential energy surface (PES), where the energies of planar and buckled geometries are two nearby local minima. Five buckled structures, namely AFM *h*-ScO, NM *h*-TiO, FM *h*-VO, AFM1 *sq*-CrO, and AFM *h*-Ti₂O₃, were found to be structural and magnetic ground-state configurations, with buckling ranging from 0.14 to 0.71 Å (see Table 1). The reason for the energetic stability of these buckled 2D phases may be the pseudo-Jahn–Teller effect⁵² (PJTE), which states that if the excited-state geometry of a polyatomic system is very close in energy (nearly degenerate) to its ground-state geometry, then these two states will be mixed via vibronic coupling, resulting in a symmetrybreaking distortion which stabilizes the system. The PJTE has been commonly seen in other 2D systems, such as group-IV^{53,54} and group-V⁵⁵ nanosheets.

Energetic stability with respect to bulk TMOs

The stability of the h-MO, sq-MO, h-M₂O₃, and sq-MO₂ phases with respect to their 3D counterparts is evaluated by comparing their total energies per atom directly:

$$\Delta E = E(2D) - E(3D). \tag{3}$$

This simple formula has been used to evaluate the stability of cadmium chalcogenide monolayers⁵⁶ and other 2D materials⁵⁷ with respect to their bulk counterparts. A negative ΔE means that a 2D monolayer is energetically more favorable than its corresponding 3D bulk phase, indicating that a material would grow into a 2D slab rather than into a 3D nanocrystal. Furthermore, Singh et al. calculated the formation energies of a range of 2D ML materials, and came to the conclusion that all 2D materials that have been synthesized experimentally in the free-standing form, have a ΔE smaller than 200 meV/atom,⁵⁸ suggesting that this could be a first selection criterion for assessing stability.

When choosing the 2D TMOs' corresponding 3D phases, in addition to having the same stoichiometry as the 2D phases, two criteria were applied to make the chosen bulk structures as relevant as possible: (1) bulk structures that can be built up directly from the 2D counterparts are preferred. This is the case for the 3D *rs*-MO and *wz*-MO phases, which can be built up directly from the 2D *sq*-MO and *h*-MO phases. (2) Bulk structures that are relatively stable at low temperatures are preferred. This is to avoid possible structural and magnetic phase transitions at higher temperatures; and it makes more sense to compare lower energy structures rather than higher-energy structures with the 2D phases, which were calculated at 0 K.

The phases that were selected as the 2D TMOs' corresponding bulk phases are listed in Table S2. Five bulk phases that have not been experimentally realized yet and remain hypothetical were still chosen because of their close similarity to the 2D TMO phases. The structural and magnetic ground states for these phases, i.e., wz-ScO, wz-TiO, wz-VO, wz-CrO, and rs-ScO, were calculated and can be found in Table S2. The magnetic ground state for rs-CrO and bixbyite Sc₂O₃, whose magnetic orderings were not found in literature, was investigated as well. No bulk phase was chosen for ScO₂: ScO₂ has neither been experimentally realized nor theoretically predicted, and there is not a clear structural connection between the 3D rutile or anatase MO₂ phase and the 2D sq-MO₂ phase. For readers' information, all bond lengths and coordination numbers for the h-MO, sq-MO, h-M₂O₃, and sq-MO₂ phases, and their corresponding 3D phases are given in Table S2 as well.

Table 2 lists the calculated values for ΔE . For all 2D TMOs except three, the calculations resulted in a ΔE larger than 200 meV/atom. The three 2D TMOs that do meet the 200 meV/atom criterion, however, were found to even have a negative ΔE . These phases are 2D *sq*-TiO, *sq*-CrO, and *sq*-MnO, and that they are energetically more favorable than 3D ϵ -TiO, *rs*-CrO, and *rs*-MnO, indicates that they may be synthesized as stable free-standing slabs. To further understand the underlying reason for the energetic stability of these *sq*-MO phases over their corresponding bulk phases, we compared the orbital-resolved density-of-states (DOS) of these 2D TMOs to the DOS of their 3D counterparts. This analysis is included as Supplementary Discussion in the SI.

Thermal stability of the 2D TMO phases

The thermal stability of the 20 structural and magnetic groundstate configurations was examined by means of AIMD simulation at temperatures of 300 and 700 K, and subsequent 0 K relaxation of AIMD output configurations to see if the structures had remained topologically equivalent. For the structures that did not show bond breaking during AIMD simulation, average bond lengths are listed in Table 3. Graphs showing minimum, average, and maximum bond lengths during 300 and 700-K AIMD simulation can be found in Fig. S1. All 300 and 700-K AIMD output configurations, and all post-AIMD relaxed structures obtained from the 300-K AIMD output configurations, can be found in Fig. S1 as well. Figure 3 shows a selection, and includes for a number of structures perspective views and post-AIMD relaxed structures obtained from 700-K AIMD output configurations, not shown in Fig. S1.

For the sq-MO phases, the AIMD and post-AIMD simulations indicate that planar sq-TiO and sq-VO are thermally stable structures. These phases show nice undulation during 300 and 700 K AIMD simulation, bond lengths increasing with increasing temperature and layers exhibiting out-of-plane displacement of atoms and wave-like vibration of the entire structure, but clearly remain the same: the 0 K relaxed AIMD output configurations for these phases are very smoothly corrugated 2D square structures that are energetically as favorable as their planar counterparts. Planar sq-ScO and mono-sided buckled sq-CrO were found to be thermally unstable: post-AIMD relaxation for these phases resulted in energetically more favorable double-sided oxygen terminated (and for sq-ScO: highly corrugated) structures that are no longer atomically thin (see Fig. 3c). Metastable sq-MnO is thermally stable at 300 K and thermally unstable at 700 K: at 300 K, this structure remained the same, its AIMD output configuration at 0 K relaxing into a very smoothly corrugated square structure (see Fig. 3b, d) that is energetically as stable as planar sq-MnO; at 700 K, sq-MnO transformed into a hexagonal phase (see Fig. 3a). The 2D square MnO→2D hexagonal MnO phase transition will be further described below.

The planar sq-MO₂ phases are relatively open structures, the number of bonds per atom being two for O and four for TM atoms

Table 2. respect			the 2D TMO phas 10s listed in Table	
тм	h-MO	sq-MO	h-M ₂ O ₃	sq-MO ₂
Sc	0.48	0.43	0.86	_
Ті	0.51	-1.60	0.60	0.93
v	0.41	0.22	0.47	0.70
Cr	0.27	-0.02	0.59	0.57
Mn	0.29	-0.26	0.44	0.65

in sq-MO₂, versus four bonds for each atom in sq-MO. An open structure could cause a layer to rupture more easily; or to be more resilient to stress, due to increased flexibility. For all sq-MO₂ phases except sq-ScO₂, it has resulted in the former: during 300-K simulation, sq-TiO₂, sq-VO₂, sq-CrO₂, and sq-MnO₂, were all found to disintegrate and partly restructure into phases that are no longer atomically thin. sq-ScO₂, however, was found to retain structural integrity and square character during 300 K simulation (bond lengths increasing by ~3%, though); and after 700-K simulation, sq-ScO₂ showed only the onset of the disintegration and partial restructuring, which for the other sq-MO₂ phases, was already and to a larger extent observed during 300 K simulation (see Fig. 3e-i). Relaxing at 0 K again, the sq-ScO₂ 300 -K AIMD output configuration was found to restructure into a phase that is no longer square and no longer atomically thin (see Fig. S1), indicating that sq-ScO₂ too is thermally unstable.

For the *h*-M₂O₃ phases, the AIMD simulations and post-AIMD relaxations show that planar $h-Sc_2O_3$ and $h-V_2O_3$ are thermally stable: the 0-K relaxed output configurations for these phases are very smoothly corrugated atomically thin 2D structures (see Fig. 3i) that are energetically as favorable as their planar counterparts. The buckled h-Ti₂O₃ and planar h-Cr₂O₃ phases were found to be unstable; during post-AIMD relaxation, the supercells for these phases restructured into phases that are no longer atomically thin (see Fig. 3I). The h-Mn₂O₃ phase is thermally stable. Both the 300 and 700 -K AIMD output configurations for this phase were found to restructure into arrangements of smoothly tilted hexagons (see Fig. 3k, m) that are only slightly more favorable than planar h- Mn_2O_3 , by 1 meV/atom, which is at the accuracy limit of the calculations. This indicates that the *h*-Mn₂O₃ structure preserves its hexagonal and atomically thin 2D character, i.e., is thermally stable at 300 and 700 K, but will exist as a smoothly corrugated structure rather than as a planar structure.

The *h*-MO phases were all found to be thermally unstable. Buckled *h*-ScO and *h*-TiO retained structural integrity during AIMD simulation; at 0 K, their AIMD output configurations relaxed, however, into structures that are no longer atomically thin (see Fig. 3n) and energetically more favorable than buckled *h*-ScO and *h*-TiO. Buckled *h*-VO, and planar *h*-CrO and *h*-MnO, also retained structural integrity during AIMD simulation and subsequently relaxed into energetically more favorable configurations; the AIMD output configurations for these three phases, however, were found to have restructured upon post-AIMD relaxation into atomically thin supercells of hexagonal "*t*-MO" phases displaying trigonal rotation symmetry (see Fig. 3o–q) that are thermally stable. (These three *t*-MO phases will be further discussed below.)

In summary (see also Table 1), of the 20 *h*-MO, *sq*-MO, *h*-M₂O₃, and *sq*-MO₂ structural and magnetic ground-state configurations, 5 were found to be thermally stable, 1 to be thermally stable at 300 K and thermally unstable at 700 K, and 14 to be unstable. The unstable

Table 3. Average bond lengths in Å for the structures that showed no bond breaking during AIMD simulation, at 300 and 700 K; showing within parentheses the increase in percent relative to the 0 K value prior to simulation

	h-MO		sq-MO		h-M ₂ O ₃		sq-MO ₂	
	300 K	700 K	300 K	700 K	300 K	700 K	300 K	700 K
Sc	2.039 (+1.1%)	2.050 (+1.6%)	2.138 (+1.3%)	2.153 (+2.0%)	1.941 (+1.1%)	1.958 (+2.0%)	2.069 (+3.3%)	_
Ti	1.959 (+2.0%)	1.970 (+2.6%)	2.039 (+1.0%)	2.059 (+2.0%)	1.847 (+1.1%)	1.862 (+2.0%)	_	_
v	1.916 (+0.4%)	1.934 (+1.4%)	2.002 (+0.8%)	2.018 (+1.6%)	1.804 (+1.0%)	1.818 (+1.9%)	_	_
Cr	1.921 (+2.3%)	1.931 (+2.7%)	2.007 (+1.0%)	2.019 (+1.5%)	1.796 (+1.3%)	1.813 (+2.3%)	_	_
Mn	1.972 (+1.9%)	1.982 (+2.4%)	2.059 (+1.5%)	_	1.805 (+1.5%)	1.825 (+2.6%)	_	_

Minimum and maximum bond lengths during simulation, and 0-K bond lengths prior to simulation, can be found in Table S3. Bond length graphs are shown in Fig. S1. That bond lengths can be evaluated does not necessarily imply that a structure is thermally stable: a number of structures for which bond lengths are listed in this Table, were found to restructure into energetically more favorable configurations, and as such are considered to be thermally unstable

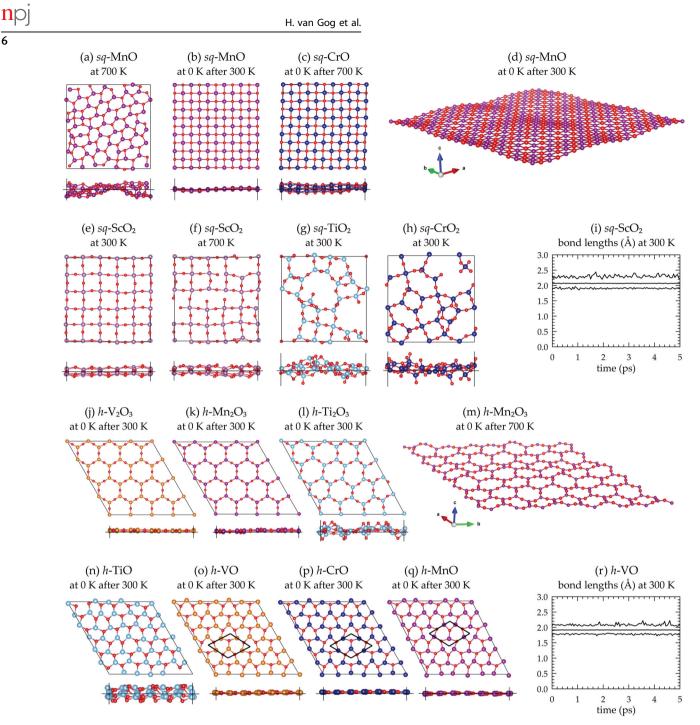


Fig. 3 AIMD simulation output configurations ("at 300 K" and "at 700 K"), post-AIMD relaxation output configurations ("at 0 K after 300 K" and "at 0 K after 700 K"), and bond length graphs, for selected structures. Phase designations refer to 0-K input configurations. Black solid lines within the post-AIMD relaxed supercells for *h*-VO, *h*-CrO, and *h*-MnO, indicate *t*-MO unit cells. Output configurations and bond length graphs for all structures can be found in Fig. S1

configurations include all buckled structures, all sq-MO₂ phases, and all h-MO phases. From the AIMD and post-AIMD simulations, another 2D hexagonal ("t-MO") phase was found, however, that is thermally stable for V, Cr, and Mn. This brings the total number of 2D TMOs identified in this study that may be realized as free-standing nanosheets to nine. These phases are: sq-TiO, sq-VO, and sq-MnO; h-Sc₂O₃, h-V₂O₃, and h-Mn₂O₃; and t-VO, t-CrO, and t-MnO.

2D square MnO \rightarrow 2D hexagonal MnO phase transition The *sq*-MnO phase was found to be metastable, hexagonal *h*-MnO and *t*-MnO being more favorable than square *sq*-MnO by 39 and 110 meV/atom, respectively (see the sections on the energetic stability of the 2D TMO phases above and below). During 700 K AIMD simulation, *sq*-MnO was found to transform into a hexagonal phase (see Fig. 3a); this transformation already starts during equilibration. The mechanism of this 2D to 2D phase transformation consists of bond breaking resulting in rectangles with four atoms at the corners and two atoms halfway the long edges, and subsequent relaxation of these rectangles into hexagons (see Fig. 4; Supplementary Video 1). Already at the end of the 2 ps equilibration, the honeycomb phase has become the dominant phase, spanning the entire width of the simulation cell, while smaller patches with square MnO remain. A rough estimate of

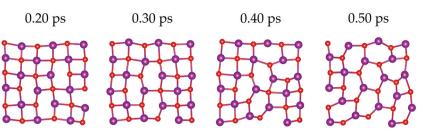


Fig. 4 Snapshots of the 700 K equilibration of sq-MnO, illustrating the mechanism of transformation of 2D square sq-MnO into 2D hexagonal t-MnO. The transformation is also shown in Supplementary Video 1

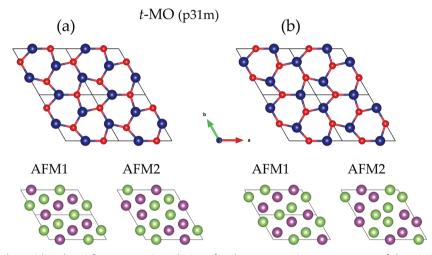


Fig. 5 Input geometry and considered anti-ferromagnetic orderings for the two atomic arrangements of the t-MO phase. Dark blue and red spheres denote TM and O atoms, respectively. Light green and purple spheres denote spin-up and spin-down TM atoms, respectively. Boundaries of simulation and unit cells are indicated with black solid lines

400 K for the transformation temperature was obtained: continuing the 300 -K simulation for another 50 ps, still no transformation of sq-MnO was observed; while during a 10 ps simulation at 400 K after initial 2 -ps equilibration, transformation into hexagonal MnO did take place.

Energetic stability of the t-MO phases

The h-VO, h-CrO, and h-MnO AIMD output configurations were found to restructure, upon post-AIMD relaxation, into atomically thin supercells of hexagonal "t-MO" phases. As can be seen from Fig. S2, the post-AIMD relaxed h-MnO supercell has an antiferromagnetic ordering corresponding to the AFM ordering found for *h*-MnO. The post-AIMD-relaxed *h*-VO output configuration was found to be FM ordered, just like *h*-VO, with the exception of one small domain. The post-AIMD relaxation output configuration obtained for anti-ferromagnetic h-CrO was found to have antiferromagnetic ordering of a type, however, that was not considered for the *h*-MOs.

To check the structural and magnetic ground state of the three t-MOs, total-energy calculations were performed, considering NM, FM, AFM1, and AFM2 magnetic ordering, planar and buckled input configurations, and two types of atomic arrangements (labeled "(a)" and "(b)" as in Fig. 5) for each of the *t*-MOs, bringing the total number of input configurations considered for the *t*-MO phases to 48 (and the total number of input configurations considered in our study to 198). Unit cell geometries and consid are illustrated in Fig. 5. Note that the t-MO reconstruction of the *h*-MO phase, lowering for perfect h-MO to p31m symmetry for perf

The results of the total-energy calculatio Table S1. Structural parameters for the structural and magnetic ground-state configurations are listed in Table 4. The ground-state

idered AFM orderings	tion for t-MnO was found to be an atomic
) phase is a $\sqrt{3} \times \sqrt{3}$	depicted in Fig. 5b. t-CrO and t-MnO are plana
the p3m1 symmetry	lowest-energy configuration for t-VO, however, v
rfect t-MO.	buckled phase, of a type where only one-thir
ons can be found in	atoms (i.e., the oxygen atoms at the $x = y$ mirror

Table 4. PBE-calculated lattice parameters and atom coordinates forthe <i>t</i> -MO phases								
	t-VO	t-CrO	t-MnO					
a (Å)	5.423	5.435	5.710					
b (Å)	_	_	5.447					
γ (°)	119.5	119.8	118.5					
(<i>x</i> , <i>y</i>)M1	(0.388, 0.004; -0.02)	(0.386, 0.001)	(0.716, 0.999)					
(<i>x</i> , <i>y</i>)M2	(0.004, 0.388; -0.02)	(0.001, 0.386)	(0.000, 0.730)					
(<i>x</i> , <i>y</i>)M3	(0.608, 0.608; 0.02)	(0.614, 0.614)	(0.284, 0.282)					
(<i>x</i> , <i>y</i>)O1	(0.724, 0.999; -0.14)	(0.718, 0.997)	(0.369, 0.989)					
(<i>x</i> , <i>y</i>)O2	(0.999, 0.724; -0.14)	(0.997, 0.718)	(0.000, 0.379)					
(<i>x</i> , <i>y</i>)O3	(0.631, 0.620)							

To make comparison between the *t*-MO phases easier and to emphasize their near-trigonal rotation symmetry, the structure parameters are of primitive unit cells, given in P1 and p1 symmetry with the atoms at Wyckoff 1a positions. Note that because of its monolayer character, the zcoordinates for t-VO are absolute coordinates in Å, whereas x and y are relative coordinates

configurations for t-VO and t-CrO were found to be atomic arrangements as depicted in Fig. 5a; the ground-state configuraarrangement as ar structures. The was found to be a ird of the oxygen plane) are located at the buckling distance of 0.44 Å (see Table 4; Fig. S2). The t-MnO phase was found to have an AFM1 ordering corresponding to its *h*-MO precursor's magnetic ground state. *t*-CrO was found to have an anti-ferromagnetic ordering of a type (AFM2) that was not considered for the *h*-MOs (*h*-CrO has an anti-ferromagnetic ordering as well, of the AFM1 type, however). *t*-VO is also AFM2 ordered; in contrast to its *h*-VO precursor, which has FM ordering.

Comparing the formation energies of the *t*-MO phases and their corresponding *h*-MO and *sq*-MO phases, it can be observed from Table 1 and Fig. 2, that just like their corresponding *h*-MO phases, *t*-VO and *t*-CrO are energetically less favorable, and *t*-MnO is energetically more favorable, than the corresponding *sq*-MO phase. *t*-VO, *t*-CrO, and *t*-MnO, are all energetically more favorable than their corresponding *h*-MO phases, by 74, 155, and 71 meV/ atom, respectively. The (latter) energetic stability of the *t*-MO phases over the *h*-MO phases may be due to the PJTE (for a brief discussion of the PJTE: see the section on the energetic stability of the *h*-MO, *sq*-MO, *h*-M₂O₃, and *sq*-MO₂ phases above).

Discussion of stability and methodology

The AIMD simulations and post-AIMD relaxations indicate that nine of the 2D TMO structural and magnetic ground-state configurations identified in this study are thermally stable (see Table 1). These phases not only retained structural integrity, i.e., did not break at elevated temperatures, but also preserved their atomically thin, and hexagonal or square, 2D character. Our findings suggest that these structures may all be synthesized as free-standing atomically thin nanosheets showing smooth corrugation upon realization. The thermally unstable 2D TMO phases can most likely not exist as free-standing slabs. These phases, however, may be stabilized using an appropriate support: substrate stabilization of 2D materials can be as large as 8.84 kJ/ mol-atom (91.6 meV/atom);⁵⁹ and substrates can be chosen to interact with the synthesized 2D TMO thin film via only a weak van der Waals force, which would stabilize the film without much interference with its electronic structure.⁶⁰ The thermally stable phases were found to include 2D TMOs that do not meet the 200 meV criterion (see the section on the energetic stability of 2D TMO phases with respect to bulk); and sq-CrO, which does meet the 200 meV criterion, was found to be thermally unstable. This merely indicates, however, that comparing 2D TMOs with corresponding bulk phases is not that straightforward, and that for a number of 2D TMO phases perhaps the corresponding bulk phase has not been chosen properly.

The AIMD as performed, imposing periodic boundary conditions on the structures, tends to overestimate structural stability: lateral periodicity mitigates out-of-plane movements of the atoms and artificially enhances structural integrity, and may result in structural integrity for structures that would otherwise curl up or even disintegrate. Furthermore, the results presented here are the outcome of 7 ps simulations; prolonged simulations might yield different results with respect to the stability of the structures. We are confident about our approach consisting of AIMD simulation and subsequent relaxation of AIMD output configurations, though, because of the proper prediction of $h-V_2O_3$: of all the structural and magnetic ground-state 2D TMO configurations identified in this study, only h-V2O3 has been experimentally realized so far,^{15,16,23} and our simulations indeed resulted in thermal stability for this phase. That $h-V_2O_3$ has been realized in a buckled form while our simulations indicate that the planar configuration is the energetically more favorable configuration for this phase, does not detract from this: our calculations suggest that $h-V_2O_3$ may be realized as free-standing smoothly corrugated planar nanosheets; the reported buckling of $h-V_2O_3$ may be due to its realization on a Pd(111) support.

Our method of AIMD simulation and subsequent relaxation of AIMD output configurations yields more reliable stability results than approaches that depend on AIMD simulation alone: Of all the 2D TMO phases identified in this study, other than $h-V_2O_3$, only $h-V_2O_3$.

MnO has been found reported on before, in a study that based on AIMD simulation only and finding no structure destruction concluded *h*-MnO to be thermally stable up to temperatures of at least 500 K.²⁸ Our method, however, indicates that *h*-MnO is thermally unstable at 300 K, already, and that any future experimental realization of an atomically thin 2D hexagonal MnO will more likely be of MnO in the *t*-MO configuration. Besides *h*-MnO, 15 more of the *h*-MO, *sq*-MO, *h*-M₂O₃, and *sq*-MO₂ groundstate configurations were found to show no structure destruction during 300 K AIMD simulation; only six of these, however, were also found to be thermally stable at 300 K. The discovery of the thermally stable *t*-MO phases in addition shows that post-AIMD relaxation of AIMD output configurations may also reveal new phases that otherwise would have remained undisclosed.

Electronic and magnetic properties of the 2D TMOs

To further investigate the magnetic properties of the 2D TMOs and to disclose their electronic structure, we calculated the electronic band structures and determined the bandgaps and atomic magnetic moments and Bader charges of the magnetic ground-state configurations, using the PBE and HSE06 functionals. A summary of the HSE-obtained results can be found in Table 1 and Fig. 6. Table S4 and Fig. S3 include the PBE results and orbital-resolved band structures.

From the comparison made in Table S4 and Fig. S3 between PBE and HSE-calculated electronic and magnetic properties, it can be seen that, in terms of an enlarged bandgap, HSE gives a better description of the strongly correlated d electrons than PBE. In particular, HSE opened bandgaps for sq-MnO, h-Cr₂O₃, sq-TiO₂, sq-VO₂, and sq-MnO₂, making these phases no longer metals as predicted by PBE, but semiconductors, semimetals, and halfmetals instead. The qualitative discrepancy between PBE and HSE occurs only when describing the electronic ground state of the 2D TMOs. For the magnetic ground state, the two approaches (i.e., PBE and HSE band-structure calculation of an identical PBE-relaxed atomic configuration) always give the same results. Regarding the Bader analysis: HSE always gives values for the atomic partial charges that are slightly larger (by 0.1-0.3e) than PBE-calculated values. Similar to the electronic charges, the magnetic moments on the atoms (that were obtained by evaluating spin-up and spindown electron densities inside Bader volumes) were also found to be slightly higher (by 0.2–0.5 μ_B) when using HSE in comparison with PBE-calculated values. Our results therefore suggest that for obtaining a gualitative description of the magnetism and charge distribution of the 2D TMOs, DFT calculations suffice. Gaining a qualitatively correct picture of the electronic structure, however, requires more advanced methods such as using the hybrid HSE06 functional that incorporates a contribution of HF exchange. The effect of this HF correction on a particular TMO phases depends on the elements, their valence states, the levels of the atomic orbitals, as well as on the crystal geometry. The general trend, however, is that the HF correction lowers the energy levels of occupied states and lifts the levels of unoccupied states, and has a strong impact on the energy levels of the itinerant d electrons. The partial atomic charges and the magnetic moments on the atoms are determined from (total, and spin-up and spin-down) electron densities inside Bader atomic volumes. As the main effect of the HF correction is on the eigenvalues of the states rather than on the electron-density distribution, the partial charges and the local magnetic moments obtained from HSE calculations are guite similar to those obtained by PBE calculations (and result in guite similar structural properties as well), while the impact on the electronic band structure is in general very significant.

Using the hybrid HSE06 functional, we found a rich variety of electronic properties among the 2D TMO phases. As can be seen from Table 1 and Fig. 6, of the 23 ground-state configurations, fifteen structures are predicted to be semiconductors and three to

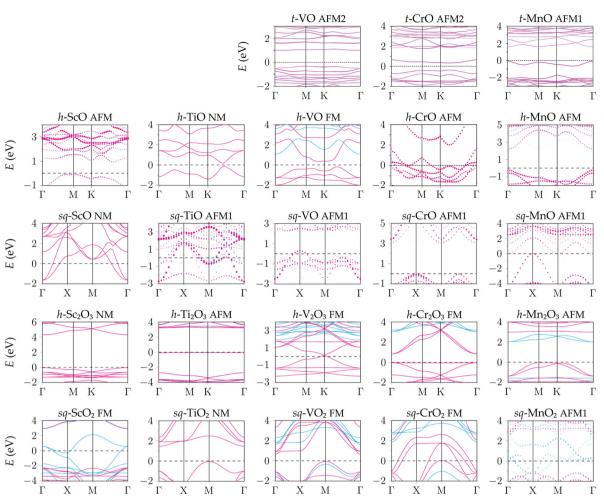


Fig. 6 HSE-calculated electronic band structures of the 2D TMO magnetic ground-state configurations. The Fermi level is set to zero and indicated with a dashed line. For spin-polarized configurations, spin-down bands are shown in cyan when spin-up and spin-down bands do not overlap

be metals with various magnetic orderings; sq-MnO was found to be an anti-ferromagnetic semimetal; and h-V₂O₃, sq-ScO₂, sq-CrO₂, and sq-MnO₂, are half-metals that are either ferromagnets (h-V₂O₃, sq-ScO₂, and sq-CrO₂) or antiferromagnets (sq-MnO₂). The magnetic moments on the atoms were evaluated from the spinup and spin-down electron densities within Bader atomic volumes, and vary up to values of 4.6 μ_B for the transition metal atoms. Besides the four half-metals, four semiconductors were also found to have non-overlapping spin-up and spin-down bands and different bandgaps for spin-up and spin-down electrons; these semiconductors are: ferromagnetic h-VO, h-Cr₂O₃, and sq- VO_2 , and ferrimagnetic *h*-Mn₂O₃. For the ferromagnetic half-metal h-V₂O₃, and for the non-magnetic semiconductor h-TiO, Diraccone-like shaped dispersion curves were found. All these remarkable properties indicate that the 2D TMOs are bright candidates for electronic, magnetic, and spintronic applications. Exceptionally remarkable in this regard is $h-V_2O_3$, that was found to be thermally stable up to temperatures of 700 K, and combines magnetism and a massless behavior of spin-up electrons along with a wide bandgap of 6.2 eV for spin-down electrons.

To understand the half-metallic property of h-V₂O₃, sq-ScO₂, sq-CrO₂, and sq-MnO₂ in more detail, we performed a spinpolarization analysis of their density-of-states (DOS) at the Fermi level. The spin polarization P is defined as

$$P = \frac{D \uparrow (E_F) - D \downarrow (E_F)}{D \uparrow (E_F) + D \downarrow (E_F)},$$
(4)

where $D(E_F)$ is the magnitude of the DOS at the Fermi level for spin-up (\uparrow) and spin-down (\downarrow) states. The spin polarization serves as a criterion to determine the extent of spin polarization, hence the degree of half-metallicity, of a material.^{61–63} $P = \pm 1$ indicates that a system is half-metallic, whereas P = 0 shows that a system has an equal amount of spin-up and spin-down states at the Fermi level. DOS analysis indeed showed that for $h-V_2O_3$, sq-ScO₂, sq-CrO₂, and sq-MnO₂, the spin polarization is equal to the ideal value of 1.

We have presented the results from our systematic firstprinciples study of atomically thin 2D TMOs. For the five early 3d transition metals Sc, Ti, V, Cr, and Mn, four different 2D TMO phases were designed, namely *h*-MO, *sq*-MO, *h*-M₂O₃, and *sq*-MO₂, and we investigated their structural and magnetic ground-state configurations, considering various magnetic orderings and planar and mono-sided buckled configurations. The 20 structural and magnetic ground-state configurations for these phases were all found to have negative formation energies with respect to the elemental transition metal bulk phases and the paramagnetic oxygen molecule. The concave-hull plot of these formation energies indicates that all the 2D TMOs are energetically stable at their respective stoichiometries.

AIMD simulation at temperatures of 300 and 700 K, and subsequent 0 K relaxation of AIMD output configurations, were carried out to examine the thermal stability of the 2D TMO phases. Of the 20 *h*-MO, *sq*-MO, *h*-M₂O₃, and *sq*-MO₂ ground-state configurations, 5 were found to be thermally stable; metastable

 $\frac{np}{10}$

sq-MnO was found to be thermally stable at 300 K and thermally unstable at 700 K, 2D square sq-MnO transforming into a 2D hexagonal MnO phase at temperatures of 400 K and higher; and 14 configurations were found to be thermally unstable. The unstable configurations include all sq-MO₂ phases, and all h-MO phases. The post-AIMD relaxations revealed a 2D hexagonal t-MO phase that is a $\sqrt{3} \times \sqrt{3}$ reconstruction of the *h*-MO phase; however, that is thermally stable for V, Cr, and Mn. This brings the total number of 2D TMOs identified in this study that may be realized as free-standing nanosheets to nine. These phases are: sq-TiO, sq-VO, and sq-MnO; h-Sc₂O₃, h-V₂O₃, and h-Mn₂O₃; and t-VO, t-CrO, and t-MnO; of all these, only h-V₂O₃ has been experimentally realized so far. Our method of AIMD simulation and subsequent relaxation of AIMD output configurations vields more reliable results than approaches that depend on AIMD simulation alone: Of all the 2D TMO phases identified in this study, other than h-V₂O₃, only h-MnO was found reported on before, in a study that, based on AIMD simulation only, predicts h-MnO to be thermally stable. Our method, however, indicates that h-MnO is thermally unstable, and that any future experimental realization of atomically thin 2D hexagonal MnO will more likely be of MnO in the t-MO phase. Our findings furthermore suggest that all thermally stable phases will show smooth corrugation upon realization. The thermally unstable 2D TMO phases can most likely not exist as free-standing slabs but may very well be stabilized when using an appropriate support.

A rich variety of electronic and magnetic properties was discovered for the 2D TMOs identified in this study, using the HSE06 hybrid functional. In particular, sq-MnO is predicted to be a semimetal with anti-ferromagnetic ordering; h-V₂O₃, sq-ScO₂, and sq-CrO₂ were found to be ferromagnetic half-metals, while sq-MnO₂ is an anti-ferromagnetic half-metal; and Dirac-cone-like bands were found for the non-magnetic semiconductor h-TiO and the ferromagnetic half-metal $h-V_2O_3$. The magnetic moments on the transition metal atoms vary up to values of 4.6 μ_B . All these remarkable properties indicate that the 2D TMOs are bright candidates for electronic, magnetic, and spintronic applications. Exceptionally remarkable in this regard is planar $h-V_2O_3$, that combines magnetism and a massless behavior of spin-up electrons along with a wide bandgap of 6.2 eV for spin-down electrons, and was found to be thermally stable up to temperatures of 700 K. The hybrid HSE06 functional gives a scenario of the electronic properties for the 2D TMOs that significantly differs from the one obtained using DFT: in general, HSE successfully opened up a bandgap for materials which were predicted to be metals by traditional DFT. This shows the necessity of using more advanced electronic structure methods to treat these strongly correlated systems correctly.

The current study assessed the stability of the 2D TMO phases and highlighted their remarkable electronic and magnetic properties, presenting an opportunity for experimentalists of targeted synthesis of these novel 2D TMO materials, which hold great potential for electronic and spintronic applications. For future work, a thorough study of the growth process and substrate effects is crucial to the successful realization of these 2D nanomaterials. Furthermore, a comprehensive study of point defects such as monovacancies, interstitials, antisites, and dopants, and the mutual interaction between these different kinds of defects, will open up the possibility of successful defect engineering in these 2D TMOs.

METHODS

Total-energy calculations

The DFT calculations were performed using the first principles' Vienna Ab initio Simulation Package^{31–33} (VASP). Within the projector-augmented wave (PAW) framework,^{34,35} the exchange and correlation energy terms were described using the PBE generalized gradient approximation (GGA)

functional.^{36,37} The cutoff energy of the wave functions was set to 700 eV, and the cutoff energy of the augmentation functions to 850 eV. All calculations were carried out using conventional unit cells or $2\times$ 2 supercells, dependent on the input configurations' magnetic ordering; except for *h*-MO, which requires a 3×3 supercell to accommodate the FiM ordering. Figure 1 illustrates the considered anti-ferromagnetic orderings for the four 2D TMO input geometries and their corresponding simulation cell sizes. Different starting values for the magnitudes of the magnetic moments on the TM atoms were considered as well, with input values of $0.0 \mu_B$ for non-magnetic calculations, and input values of 1.0, 2.0, 3.0, 4.0, and $5.0 \mu_B$ for magnetic calculations. To avoid spurious interactions between layers, a vacuum slab that is larger than 20 Å was included in all simulation cells. The electronic wave functions were sampled on grids in inverse proportion to simulation cell size, using a $16 \times 16 \times 17$ -centered kmesh for the conventional unit cells. Cutoff energies, k-mesh, and height of the vacuum slab were all tested to ensure energy convergence within 1 meV/atom. The structures were structurally and magnetically optimized, keeping the cell volume fixed to avoid collapse of the vacuum slab separating the model layers, and using energy and atomic force convergence criteria of 10^{-6} eV and 10^{-2} eV/Å, respectively. The calculations are valid for a temperature of 0 K and a pressure of 0 GPa. Zero-point vibration contributions have been neglected. Non-collinear magnetism and spin-orbit coupling have not been considered.

Band-structure analysis

The electronic structure of the PBE-relaxed configurations was evaluated using the hybrid HSE06 functional.²⁶ The fraction of HF exchange for these calculations was set to 0.25, resulting in a $\frac{1}{4}$: $\frac{3}{4}$ hybridization of the HF exchange energy and PBE exchange energy, and the cutoff energy of the wave functions and the augmentation functions to 500 and 600 eV, respectively. Electronic band structures obtained from supercells were unfolded using the method by Tomić et al.⁶⁴

AIMD simulations

The AIMD simulations were performed using 72-atom 6×6 supercells for the *h*-MO phases, 100-atom 5×5 supercells for the *sq*-MO phases, 80-atom 4×4 supercells for the *h*-M₂O₃ phases, and 72-atom 5×5 supercells for each of the *sq*-MO₂ phases, except *sq*-MnO₂, which was simulated using a 108-atom 6×6 supercell to accommodate its magnetic ordering. For each of the structures, with a time step of 1 fs, a 2 ps initial equilibration and a 5 ps subsequent canonical (NVT) ensemble simulation using a Nosé thermostat^{49–51} was carried out, at simulation temperatures of 300 and 700 K. The AIMD calculations were performed including the Γ -point only and using a cutoff energy of 400 eV for the wave functions. After the AIMD simulations, AIMD supercell output configurations were relaxed at high accuracy at 0 K again, to see if the structures had remained topologically equivalent.

DATA AVAILABILITY

Details on the crystal structures and calculation settings are provided here and the Supplementary Information. The calculation data will be stored in the Data Repository of Utrecht University, and are available from the authors upon request.

ACKNOWLEDGEMENTS

This project is financially supported by the Dutch science foundation NWO via a VIDI grant (grant no. 723.012.006) and by the European Research Council through an ERC Consolidator Grant (grant no. 683076). This work was carried out on the Dutch national e-infrastructure with the support of SURF Cooperative and NWO Rekentijd (grant no. 16905). Figures of atomic structural models were produced using VESTA.⁶⁵ W.-F.L. gratefully acknowledges J.-P. Chou and S. S. Gupta for valuable discussions. The authors furthermore appreciate the careful reading and helpful comments on the paper from A. van Blaaderen.

AUTHOR CONTRIBUTIONS

H.v.G. performed the AIMD and post-AIMD calculations, including the total-energy calculations and band-structure analysis for the *t*-MO phases. W.F.L., M.A.v.H., and H.v. G. performed the total-energy calculations and band-structure analysis for the *h*-MO, *sq*-MO, *h*-M₂O₃, and *sq*-MO₂ phases. H.v.G. wrote the paper, incorporating an early

version of the paper by W.F.L. All authors contributed to discussions and commented on the paper.

ADDITIONAL INFORMATION

Supplementary information accompanies the paper on the *npj 2D Materials and Applications* website (https://doi.org/10.1038/s41699-019-0100-z).

Competing interests: The authors declare no competing interests.

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